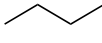
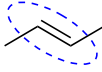


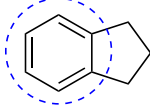
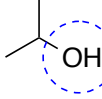
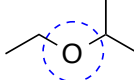
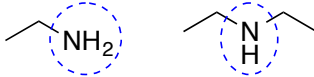
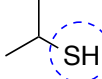
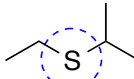
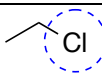

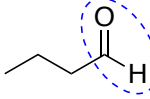

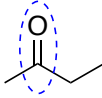

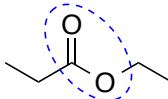

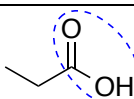

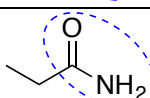

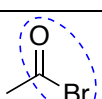


Chemistry 233
Functional Groups and Alkane Nomenclature
Organic Functional Groups

| Hydrocarbons | | |
|--|---|---|
| Name | Functional Group | Examples |
| Alkane (not really a functional group) | R-H |  |
| Alkene | R ₂ C=CR ₂ |  |
| Alkyne | RC≡CR |  |
| Arene (Aromatic) |  |  |
| Heteroatom Containing | | |
| Alcohol (hydroxy group) | R-OH |  |
| Ether | R-O-R |  |
| Amine (amino group) | R-NH ₂ R ₂ NH R ₃ N |  |
| Thiol (mercapto group) | R-SH |  |
| Sulfide | R-S-R |  |
| Halide | R-X (X = F, Cl, Br, I) |  |
| Carbonyl (C=O) Containing | | |
| Aldehyde |  |  |
| Ketone |  |  |
| Ester |  |  |
| Carboxylic Acid |  |  |
| Amide |  |  |
| Acid Halide |  |  |

Straight Chain Alkanes

| Name | # of Carbons | Condensed Structure | Skeletal Structure |
|---------|--------------|---|--------------------|
| Methane | 1 | CH ₄ | None |
| Ethane | 2 | CH ₃ CH ₃ | |
| Propane | 3 | CH ₃ CH ₂ CH ₃ | |
| Butane | 4 | CH ₃ (CH ₂) ₂ CH ₃ | |
| Pentane | 5 | CH ₃ (CH ₂) ₃ CH ₃ | |
| Hexane | 6 | CH ₃ (CH ₂) ₄ CH ₃ | |
| Heptane | 7 | CH ₃ (CH ₂) ₅ CH ₃ | |
| Octane | 8 | CH ₃ (CH ₂) ₆ CH ₃ | |
| Nonane | 9 | CH ₃ (CH ₂) ₇ CH ₃ | |
| Decane | 10 | CH ₃ (CH ₂) ₈ CH ₃ | |

Some Common Alkyl Substituents

R=parent chain

| | | | |
|--------------|-------------|--------------|------------------|
| $R-CH_3$ | | | |
| Methyl (-Me) | Ethyl (-Et) | Propyl (-Pr) | Isopropyl (-iPr) |

| | | | |
|-------|-----------|----------|------------|
| | | | |
| Butyl | sec-Butyl | isobutyl | tert-Butyl |

Summary of Rules for Naming Acyclic Alkanes

- Find the longest continuous carbon chain and name it as the alkane parent.**
 - If two chains of equal length are found, pick the one with the most branches as the parent.
- Number the carbon atoms in the chain.**
 - Number in the direction that gives the first substituent the lowest number.
 - If numbering from both directions gives the first substituent the same number, then number so that the second substituent has the lowest possible number.
 - If there is a numbering tie in both directions, then you should use alphabetical priority.
- Number and name substituents.**
 - Place a number in front of a substituent name followed by a dash (i.e. 2-ethyl). If there is more than one of the same substituent present, designate with a prefix (di-, tri-, tetra-, penta-){i.e. 2,4,6-triethyl}.
- List substituents in alphabetical order followed by the parent name.**
 - Prefixes such as di-, tri-, tetra-, penta-, *sec-*, and *tert-* do not count for alphabetization purposes. Iso and cyclo both count for alphabetization purposes.
 - Numbers are separated with commas. Numbers and names are separated with dashes.